

# 10.1 Periodic Trends



## PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

### Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	Aluminum
_____	14-18
_____	Transition?

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

#### Column A

- \_\_\_\_\_ 19. It exists in the periodic table.
- \_\_\_\_\_ 20. It has no line periodic table.
- \_\_\_\_\_ 21. Group 1 elements.
- \_\_\_\_\_ 22. Elements that are strong oxidizing agents.
- \_\_\_\_\_ 23. Group 18 elements.

#### Column B

- a. noble
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 24. There are two representative elements of elements.
- \_\_\_\_\_ 25. Most elements have lines in the periodic table for elements.
- \_\_\_\_\_ 26. Group 1A elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 27. Group 1A elements are the alkali metal metals.
- \_\_\_\_\_ 28. Group 1A elements are highly reactive elements in hydrogen.
- \_\_\_\_\_ 29. Group 1A elements are very reactive elements found in transition metals.
- \_\_\_\_\_ 30. Alkali metals have properties of both metals and non-metals.

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**Patricia M Nugent, Barbara A Vitale**



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