

# Nuclear Chemistry

- We hear about nuclear power and the results of the atom only when there is a Tragic accident

- Three Mile Island
- Chernobyl
- Fukushima

It is important to have a understanding of radiation and there effects

- Background Radiation
  - We are all exposed to radiation
  - Three-fourths of all exposure to radiation comes from **background radiation**.
  - Most of the remaining one-fourth comes from medical irradiation such as X-rays.
- 
- **Average origin of radiation we receive**
- We receive about 0.3 rem of radiation each year
  - 55 % from radon in indoor and outdoor air
  - 8 % from cosmic rays from outer space
  - 8 % from rocks and soil
  - 11 % from natural radioactive isotopes ( $^{14}\text{C}$ ) of elements that are present in our own bodies
  - 18 % from anthropogenic sources (mainly medical x- rays)
  - Contribution from nuclear power plants is negligible (on average)

## Radon

- Guidelines for radon levels in indoor air:
  - Health Canada set the maximum to  $200 \text{ Bq/m}^3$
  - U.S. EPA set the maximum to  $4 \text{ pCi/L} = 148 \text{ Bq/m}^3$
- Radiation Damage to Cells
  - Radiation is capable of removing electrons from cells forming ions, hence the term **ionizing radiation**.
  - Molecules can also splinter into neutral fragments called **free radicals**. Free radicals can disrupt cellular processes.

# Study Guide For Nuclear

**Raymond A. Serway, Steven Van Wyk**



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