

THE ULTIMATE GENERAL CHEMISTRY 1 STUDY GUIDE

- ❖ Matter and Measurement
 - Matter: anything that has mass and occupies space
 - Properties of matter: physical properties and chemical properties
 - States of matter: solid, liquid, gas
 - Measurements: SI Units, significant figures, dimensional analysis
- ❖ Atoms, Molecules, and Ions
 - Atomic structure: protons, neutrons, and electrons
 - Periodic table and trends: electronegativity, ionization energy, electron affinity
 - Chemical bonding: ionic bonding, covalent bonding, metallic bonding
 - Chemical formulas and equations: empirical and molecular formulas, balancing chemical equations
- ❖ Stoichiometry
 - Limiting reactants and percent yield
 - Mole concept and Avogadro's number
 - Molarity and molality
 - Gas laws: Boyle's law, Charles's law, Gay-Lussac's law, Avogadro's law, Ideal gas law
- ❖ Chemical Reactions and Equations
 - Types of chemical reactions: synthesis, decomposition, single replacement, double replacement, combustion
 - Redox reactions and balancing redox equations
 - Reaction rates: factors affecting reaction rates, rate laws, and reaction mechanisms
- ❖ Gases
 - Gas laws: Boyle's law, Charles's law, Gay-Lussac's law, Avogadro's law, Ideal gas law
 - Kinetic Molecular Theory of gases: pressure, temperature, volume, and the number of moles of gas particles
 - Natural gases: van der Waals equation

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**Colorado Agricultural and Mechanical
College. Department of
Chemistry, Paul Reheard Frey**

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Chemistry 1411 Enrique Olivas,2014-08-19 THE MAIN OBJECTIVES DURING THE DEVELOPMENT OF THIS BOOK WAS TO BETTER PREPARE STUDENTS THAT NEED TO TAKE THE FINAL COMPREHENSIVE AMERICAN CHEMICAL SOCIETY ACS EXAM AS WELL AS THOSE STUDENTS THAT SEEK ADMISSION IN MEDICAL AND PHARMACY SCHOOLS THE STUDY GUIDE DESCRIBES IN AN OUTLINE FORMAT THE MOST IMPORTANT TOPICS COVERED IN GENERAL CHEMISTRY I THE BOOK SUMMARIZES THE MAIN OBJECTIVES THAT STUDENTS ENROLLED IN THE COURSE SHOULD LEARN THE BOOK GIVES MANY SAMPLE PROBLEMS WITH STEP WISE SOLUTIONS SO THAT STUDENTS CAN FOLLOW THE MATERIAL EASIER WHEN NECESSARY MATHEMATICAL FORMULAS ARE GIVEN ALL THROUGHOUT TO FACILITATE THE SOLUTIONS TO NUMERICAL PROBLEMS THIS STUDY GUIDE CAN BE USED BY ANY COLLEGE STUDENT ENROLLED IN GEN CHEM I REGARDLESS OF THE TEXT USED ONE OF THE PITFALLS OF MOST TEXTS IS THE EXCESSIVE AMOUNT OF MATERIAL COVERED BUT FAIL TO EMPHASIZE THE MOST IMPORTANT FEATURES OF THE TOPIC COVERED

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