

BALANCING EQUATIONS CHEAT SHEET

Steps for Balancing a Chemical Equation

Step 1: Write the unbalanced chemical equation.

Step 2: Choose an element that appears in only one reactant and one product.

Step 3: Balance the element by changing the coefficient in front of the molecule containing it.

Step 4: Repeat steps 2 and 3 until the equation is balanced.

- The coefficient in front of a molecule is always written to the left of the molecule.
- The coefficient in front of a molecule is always written to the left of the molecule.

Example 1: Balancing a Chemical Equation



Table 1: Balancing

	Reactants	Products
C	2	1
H	6	2
O	2	3

Table 2: Balancing

Reactants	Products	Reactants	Products
C ₂ H ₆	2	CO ₂	2
C ₂ H ₆	6	H ₂ O	6
O ₂	2	O ₂	2

Table 3: Balancing

Reactants	Products	Reactants	Products
C ₂ H ₆	2	CO ₂	2
C ₂ H ₆	6	H ₂ O	6
O ₂	2	O ₂	2
O ₂	2	O ₂	2

Table 4: Balancing

	Reactants	Products
C	2	2
H	6	6
O	2	2

Step 5: Repeat steps 2 and 3 until the equation is balanced.

Step 6: Check the balanced chemical equation.

Step 7: Repeat steps 2 and 3 until the equation is balanced.

Step 8: Check the balanced chemical equation.



Table 5: Balancing

Reactants	Products	Reactants	Products
C ₂ H ₆	2	CO ₂	2
C ₂ H ₆	6	H ₂ O	6
O ₂	2	O ₂	2
O ₂	2	O ₂	2

Table 6: Balancing

	Reactants	Products
C	2	2
H	6	6
O	2	2

Step 9: Repeat steps 2 and 3 until the equation is balanced.



Table 7: Balancing

Reactants	Products	Reactants	Products
C ₂ H ₆	2	CO ₂	2
C ₂ H ₆	6	H ₂ O	6
O ₂	2	O ₂	2

Table 8: Balancing

Reactants	Products	Reactants	Products
C ₂ H ₆	2	CO ₂	2
C ₂ H ₆	6	H ₂ O	6
O ₂	2	O ₂	2
O ₂	2	O ₂	2

Table 9: Balancing

	Reactants	Products
C	2	2
H	6	6
O	2	2



Study Guide For Balancing Equations

Pamphlet Master



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