

Introduction to Chemistry

Section 1.1 The Stories of Two Chemicals

In your textbook, read about the ozone layer.

Use each of the boxes below just once to complete the passage.

atmosphere	oxygen gas	ozone	ozone hole
ionosphere	hydroxyls	ultraviolet radiation	

Earth's **OZ** _____ is made up of several layers. The air ionosphere makes up the lowest level. This layer is called the **OZ** _____. The next layer up is called the **OZ** _____. This layer contains a protective _____.

OH _____ layer.

Chemical reactions **OZ** _____ is caused by ultraviolet radiation in the upper part of the atmosphere. The ozone layer is layer around Earth, without ozone _____.

OH _____. Without ozone, you get more likely to get a rash or a possibly skin cancer. The thinning of the ozone layer called the **OZ** _____ is measured because without ozone all organisms on Earth are subject to harm from too much radiation.

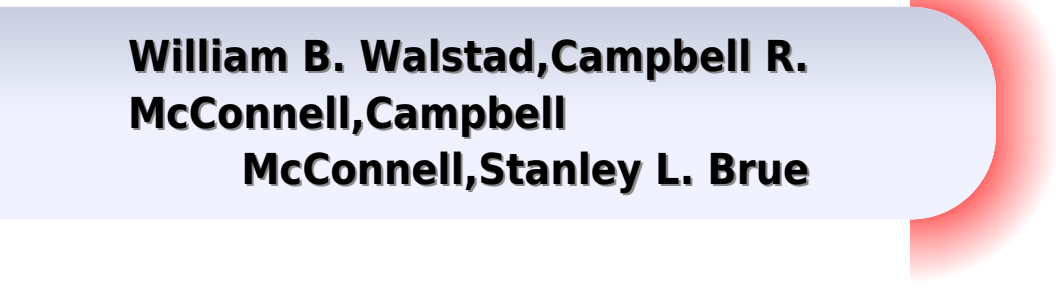
In your textbook, read about chlorofluorocarbons.

Use each statement below, with one or more _____.

- _____ **1**. CFCs is another name for a chlorofluorocarbon.
- _____ **2**. CFCs are made up of carbon, fluorine, and chlorine.
- _____ **3**. All CFCs are synthetic chemicals.
- _____ **4**. CFCs usually react slowly with other chemicals.
- _____ **5**. CFCs were developed as replacements for toxic refrigerants.

Study Guide For Content Mastery Chapter 8

**William B. Walstad, Campbell R.
McConnell, Campbell
McConnell, Stanley L. Brue**



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