

PLTW Biomedical Science Final Study Guide

What is a calorie? correct answersA unit of heat energy

With our crude calorimeters in class we burned food by lighting it on fire. What chemical reaction occurs when food is burned? How does this reaction relate to the breakdown of food for energy in our body? correct answersA combustion reaction occurs. The food and oxygen in the air combine to form carbon dioxide and water molecules and produce lots of energy. This happens in the body as well, using enzymes to catalyze all of the reactions.

If you burned a sample of food and the temperature of 100g of water in your calorimeter rose from 20°C to 36°C, how many calories were in the food sample?

Calories = (specific heat of water) x (change in temp of water) x (mass of water) Specific heat of water = 1 calorie/g°C correct answersCalories = (1 cal/g°C) (16°C) (100g) = 1600 calories

If the food sample you burned above had a mass of 10g before burning and 8g after burning, how many calories/gram did the food contain? correct answers1600 calories/2g = 800 calories/gram

We are expressing our calories calculated above as chemistry calories. How are food calories different? correct answers1 food Calorie = 1000 chemistry calories

Which macromolecules -- carbohydrates, proteins or lipids - have the highest calories/gram? correct answersLipids have the highest: 9 Calories/gram.

In our bodies the food we eat is broken down into its building blocks as it passes through our digestive system. Answer the following questions about this process:

Where does digestion begin? correct answersIn the mouth with active chewing and the enzymes in saliva.

The enzymes amylase and lipase are found in saliva. Which macromolecules do these enzymes break down? correct answersAmylase breaks down carbohydrates. Lipase breaks down lipids.

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