

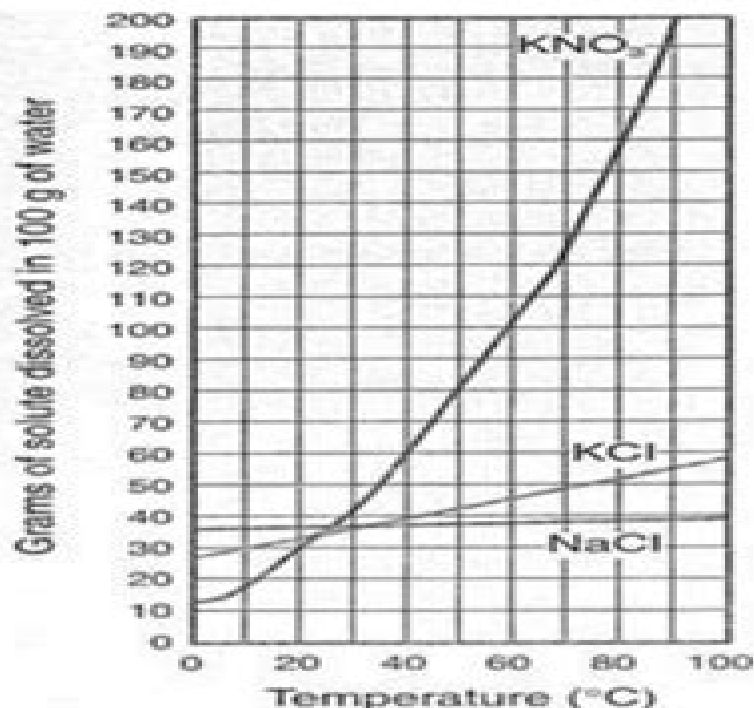
## Solubility Graph Reading

Name \_\_\_\_\_ Hr \_\_\_\_\_

Use the solubility graphs to answer the following questions.

- 1) What mass (in grams) of KCl will dissolve in 100 grams of water at 70°C?
- 2) How many more or fewer grams of NaCl will dissolve in 100 grams of water at 70°C than grams of KCl in #1?
- 3) What mass of potassium nitrate will dissolve in 100 grams of water at 70°C?
- 4) How much more potassium nitrate can be dissolved in 100 g of water that has been raised from 10°C to 50°C and originally contained the saturated amount?
- 5) 25g of potassium nitrate are dissolved in 100 grams of water and the temperature is raised to 90°C how much more chemical is required to form a saturated solution?

Effect of Temperature on the Solubility of Various Salts in Water



- 6) If the temperature of potassium chloride is dropped from 100°C to 40°C how much chemical will precipitate from solution if no supersaturated solution is formed?
- 7) Describe the relationship between temperature and solubility of a solid.

# Study Guide Solubility Graph Key

**Lawrence H. Van Vlack**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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