

MIXTURES & SOLUTIONS

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Study Guide

MIXTURES & SOLUTIONS Study Guide

Before each of the following activities, I can identify the purpose of the activity. I can distinguish between a mixture and a solution. I can describe materials that can dissolve in water. I can explain how to separate mixtures and which method to use.

Word	Definition
Mixture	
Solution	
Dissolve	
Solvent	
Solute	
Substance	

- Highlight any phrases that are describing a solution:
- Excludes only solid substances
 - Excludes liquids
 - Excludes substances in solid
 - Must not be separated
 - Must cannot be separated
 - Something dissolves

Order the mixtures below:



Do you want a substance to dissolve quicker?

- Decrease the _____ of the solvent
- _____ the substance in the solvent
- Put the solute into _____ pieces
- Use greater _____ surface area

Name 1 substance that dissolves in water:

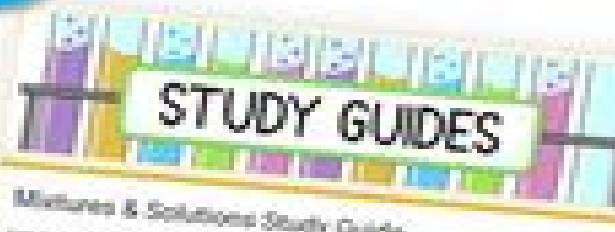
Name 1 substance that does NOT dissolve in water:

What can you suggest to happen when you mix oil and water?

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SEPARATING MIXTURES

Required materials	Describe what happens
Method name: Density Objects of different densities float	Place objects of different densities into water. Observe objects and note which objects float.
Method name: Magnet Magnetic (solid) Nonmagnetic (solid)	Separating materials from each other using a magnet.
Method name: Filter Small particles Larger particles	Separate small particles from liquids by pouring them through a filter. Observe water drops in.
Method name: Evaporation Small particles Larger particles	Separate small particles from liquids by pouring the mixture into a shallow dish. Leave the dish in the sun. Observe the liquid evaporating.
Method name: Distillation Small particles Larger particles	Separate a substance from a mixture by heating the mixture. Observe the liquid evaporating and condensing.



Study Guide Mixtures And Solutions

Susan McMurry, John McMurry



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Table of Contents Study Guide Mixtures And Solutions

1. Understanding the eBook Study Guide Mixtures And Solutions
 - The Rise of Digital Reading Study Guide Mixtures And Solutions
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Mixtures And Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide Mixtures And Solutions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Mixtures And Solutions
 - Personalized Recommendations
 - Study Guide Mixtures And Solutions User Reviews and Ratings
 - Study Guide Mixtures And Solutions and Bestseller Lists
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- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Study Guide Mixtures And Solutions
 - Highlighting and Note-Taking Study Guide Mixtures And Solutions
 - Interactive Elements Study Guide Mixtures And Solutions
- 8. Staying Engaged with Study Guide Mixtures And Solutions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide Mixtures And Solutions
- 9. Balancing eBooks and Physical Books Study Guide Mixtures And Solutions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide Mixtures And Solutions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Study Guide Mixtures And Solutions
 - Setting Reading Goals Study Guide Mixtures And Solutions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Study Guide Mixtures And Solutions
 - Fact-Checking eBook Content of Study Guide Mixtures And Solutions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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