

# Bonding, Naming, Molecular Structures Study Guide

5 pages + 5-page Answer Key

## BONDING, NAMING, AND MOLECULAR SHAPES STUDY GUIDE

1. What are valence electrons, and why are they important in chemical bonding?

2. How does the octet rule influence the bonding?

3. What distinguates ions resulting from electron transfer and the region of elements involved?

4. List the common molecular shapes and bond angles.

5. What are molecular structures, and how do they affect the properties of a substance?

6. Draw the Lewis structure for the following compounds:

7. Draw the Lewis structure for the following compounds:



8. What naming rules must be followed when naming a compound?

9. Why is it important to ensure that a compound is named correctly?

10. Name the following compounds:

NaCl	
KCl	
MgCl <sub>2</sub>	
CaCl <sub>2</sub>	
FeCl <sub>3</sub>	

11. Draw the Lewis structure for the following compounds:



12. Draw the Lewis structure for the following compounds:



Chemistry

Physical Science

13. What are the common molecular shapes and bond angles?

14. What are the common molecular shapes and bond angles?

15. Why is it important to ensure that a compound is named correctly?

16. How will drawing and identifying molecular shapes enhance your understanding of chemistry?

ANSWER KEY INCLUDED

# Study Guide Molecular Structures Answers

**Kenneth W. Raymond**



## Study Guide Molecular Structures Answers:

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