

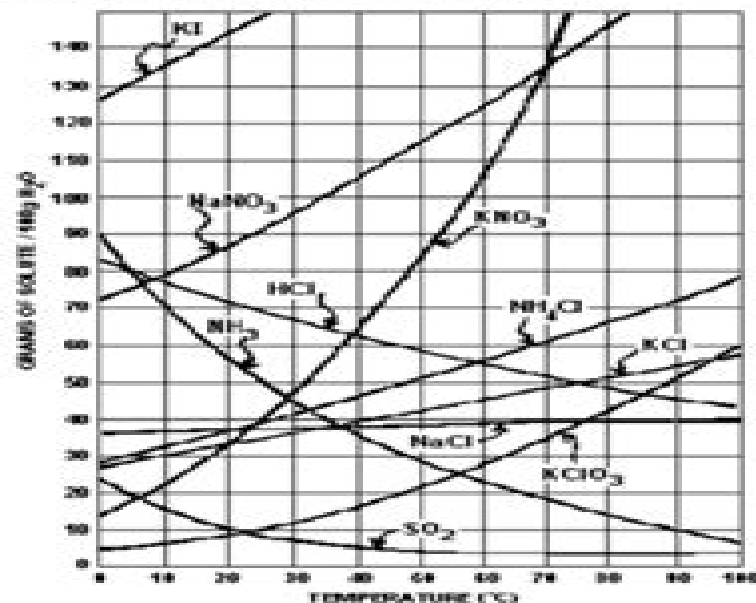
Solutions Level II

Section I: Reading Solubility Curves – Use the solubility curve below to answer the following questions.

1. What relationship exists between solubility and temperature for the ionic substances shown?
2. What relationship exists between solubility and temperature for the only gas, SO_2 , on the graph?

Indicate if the following solutions are saturated, unsaturated or supersaturated.

3. 20 grams of NH_3 completely dissolved in 100 grams of water at 80°C .
4. 80 grams of KNO_3 completely dissolved in 100 grams of water at 55°C ?
5. 112 grams of NaNO_3 completely dissolved in 100 grams water at 45°C ?
6. 10 grams of KClO_3 completely dissolved in 100 grams water at 60°C ?
7. 30g KCl in 100g water at 10°C ?



Section II: Electrolytes - Complete the following table. The first one has been done for you as an example.

	Formula?	Ionic, Acid/Base, or Covalent?	Soluble in water?	Electrolyte?
Sodium chloride	NaCl	Ionic	Yes	Yes
1. Acetic Acid				
2. Silver bromide				
3. Hydrofluoric acid				
4. Carbon dioxide				
5. Strontium hydroxide				
6. Magnesium chlorate				
7. Ammonium nitrate				

Section I: 1. Directly related; 2. Inversely related; 3. S; 4. U; 5. S; 6. U; 7. S Section II: 1. Y 2. N 3. Y 4. N 5. Y 6. Y 7. Y

Study Guide Solubility Graph Answers

**Norman E. Griswold, Henry Fuller
Holtzclaw**



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