

Stoichiometry, Limiting Reactants, Percent Yield, and Excess Reactant Remaining Worksheet

1. Aluminum reacts with oxygen to produce aluminum oxide which can be used as an adsorbent, desiccant or catalyst for organic reactions.

1 mole of aluminum (Al) is 26.982 g
1 mole of molecular oxygen (O₂) is 31.998 g
1 mole of aluminum oxide is 101.961 g

a. What is the formula of aluminum oxide? (First hint: use the criss-cross rule with aluminum ion and oxide ion) (Second hint: check your answer with the molar mass of aluminum oxide)



b. Write the balanced chemical equation for the reaction of aluminum (Al) with molecular oxygen (O₂) to make aluminum oxide.



c. How many grams of aluminum oxide could form if 82.49 g Al reacts completely?

$$82.49 \text{ g Al} \times \frac{1 \text{ mol Al}}{26.982 \text{ g Al}} \times \frac{2 \text{ mol Al}_2\text{O}_3}{4 \text{ mol Al}} \times \frac{101.961 \text{ g Al}_2\text{O}_3}{1 \text{ mol Al}_2\text{O}_3} = 155.9 \text{ g Al}_2\text{O}_3$$

d. How many grams of aluminum oxide could form if 117.65 g O₂ reacts completely?

$$117.65 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{31.998 \text{ g O}_2} \times \frac{2 \text{ mol Al}_2\text{O}_3}{3 \text{ mol O}_2} \times \frac{101.961 \text{ g Al}_2\text{O}_3}{1 \text{ mol Al}_2\text{O}_3} = 249.9 \text{ g Al}_2\text{O}_3$$

Study Guide Stoichiometry Limiting Reactant Percent Yield

Leo J. Malone

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Study Guide to Accompany Calculus for the Management, Life, and Social Sciences Clyde Metz, 1984-01-01 Study Guide to Accompany Calculus for the Management Life and Social Sciences Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook. Each chapter contains Overview, Topical Outline, Skills and Common Mistakes which are all keyed to the textbook for easy cross reference. The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms, a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook. The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples. The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter. After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter, energy, measurement, problem solving, atom, periodic table and chemical bonding. These topics are followed by discussions on writing names and formulas of compounds, chemical formulas and the mole, chemical reactions, calculations based on equations, gases and the properties of a liquid. The remaining chapters examine the solutions, acids, bases, salts, oxidation-reduction reactions, electrochemistry, chemical kinetics and equilibrium and nuclear, organic and biological chemistry. This study guide will be of great value to chemistry teachers and students.

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes. Also includes

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