

Name: Date:

Student Exploration: Limiting Reactants

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make?

2. Which ingredient limited the number of hot dogs you could make?

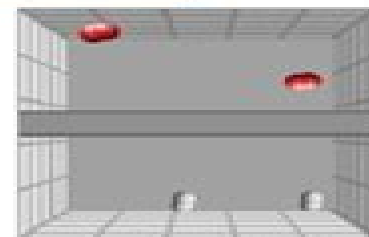
3. Which ingredient will you have leftovers of?

Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.

To begin, make sure **$\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$** is selected. The small "2" in H_2 , O_2 , and H_2O is a **subscript**. Subscripts represent the number of atoms in a **molecule**.



1. Use the sliders to set the number of **O_2 molecules** and **H_2 molecules** to two.

A. How many hydrogen molecules (H_2) are there?

Hydrogen atoms?

B. How many oxygen molecules (O_2) are there?

Oxygen atoms?

2. How many H_2O molecules do you think will form when these four molecules react?

3. Click **Play** (). How many H_2O molecules actually formed?

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Victor M. Corman



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Table of Contents Student Exploration Guide Limiting Reactants Answer Key

1. Understanding the eBook Student Exploration Guide Limiting Reactants Answer Key
 - The Rise of Digital Reading Student Exploration Guide Limiting Reactants Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Guide Limiting Reactants Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Student Exploration Guide Limiting Reactants Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Student Exploration Guide Limiting Reactants Answer Key
 - Personalized Recommendations
 - Student Exploration Guide Limiting Reactants Answer Key User Reviews and Ratings
 - Student Exploration Guide Limiting Reactants Answer Key and Bestseller Lists
5. Accessing Student Exploration Guide Limiting Reactants Answer Key Free and Paid eBooks
 - Student Exploration Guide Limiting Reactants Answer Key Public Domain eBooks
 - Student Exploration Guide Limiting Reactants Answer Key eBook Subscription Services
 - Student Exploration Guide Limiting Reactants Answer Key Budget-Friendly Options
6. Navigating Student Exploration Guide Limiting Reactants Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Student Exploration Guide Limiting Reactants Answer Key Compatibility with Devices
 - Student Exploration Guide Limiting Reactants Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Guide Limiting Reactants Answer Key
 - Highlighting and Note-Taking Student Exploration Guide Limiting Reactants Answer Key
 - Interactive Elements Student Exploration Guide Limiting Reactants Answer Key

8. Staying Engaged with Student Exploration Guide Limiting Reactants Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Guide Limiting Reactants Answer Key
9. Balancing eBooks and Physical Books Student Exploration Guide Limiting Reactants Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Student Exploration Guide Limiting Reactants Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Student Exploration Guide Limiting Reactants Answer Key
 - Setting Reading Goals Student Exploration Guide Limiting Reactants Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Student Exploration Guide Limiting Reactants Answer Key
 - Fact-Checking eBook Content of Student Exploration Guide Limiting Reactants Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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