

Chemistry Unit 3

Study Guide

Chemical Nomenclature

Chemical formula

- a. A way of presenting information about the chemical proportions of atoms that constitute a particular chemical compound or molecule, using chemical element symbols, numbers, and sometimes also other symbols

Subscript

- a. A subscript or superscript is a character that is set slightly below or above the normal line of type, respectively. It is usually smaller than the rest of the text. Subscripts appear at or below the baseline, while superscripts are above.

Ion

- a. Ion, any atom or group of atoms that bears one or more positive or negative electrical charges.

Anion

- a. A negatively charged ion.

Cation

- a. A positively charged ion.

Polyatomic ion

- a. The prefix poly- means many, so a polyatomic ion is an ion that contains more than one atom.

Understand the use of Roman Numerals in ionic compounds

- a. Roman numerals are used in naming ionic compounds when the metal cation forms more than one ion.

List the diatomic molecules

- a. Hydrogen (H₂)
- b. Nitrogen (N₂)
- c. Oxygen (O₂)
- d. Fluorine (F₂)
- e. Chlorine (Cl₂)
- f. Iodine (I₂)
- g. Bromine (Br₂)

Be able to name and write chemical formulas for ionic, covalent, and acidic compounds

- Properly use prefixes
- Properly use Roman Numerals
- Properly use endings (-ide, -ate, -ite, -ous, -ic)

Determine the type of compound made from a pair of elements or polyatomic ions

Identify the ionic charge of transition metals in compounds

- a. Because most transition metals have two valence electrons, the charge of 2+ is a very common one for their ions.

Identify compounds as ionic, covalent, or acidic

- a. Compounds containing two elements (so called binary compounds) can either have ionic or covalent bonding. If a compound is made from a metal and a nonmetal, its bonding will be ionic. If a compound is made from two non-metals, its bonding will be covalent.

Unit 3 Chemistry Study Guide Answers

Kathy Austin

Unit 3 Chemistry Study Guide Answers:

Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition Susan McMurry,2000

Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Price winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines

Resources in Education ,1980 **Basic Concepts of Chemistry, Study Guide** Leo J. Malone,2000 The text s three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems **Gage Canadian GED Study Guide Series** Kathy Austin,1998

Study Guide General Chemistry, Second Edition, Becker/Wentworth Becker,Benjamin L. Carroll,1980

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PM310, 23hp Kawasaki Wiring Diagram Gravely 990020 (001000 -) PM310, 23hp Kawasaki Wiring Diagram Exploded View parts lookup by model. Complete exploded views of all the major manufacturers. Kawasaki Barako BC 175 Electrical Wiring Update Aug 11, 2017 — If there are no problems on the wirings and connectors; 2. Check the input to the VR, there are two wires coming from the charging coils. One is ...