

Temperature Mix

Purpose

To predict the final temperature of a mixture of cups of water at different temperatures

Apparatus

3 Styrofoam cups
liter container with a wide mouth
thermometer (Celsius)
pail of cold water
pail of hot water

Discussion

If you mix a pail of cold water with a pail of hot water, the final temperature of the mixture will be between the two initial temperatures. What information would you need to predict the final temperature? You'll begin with the simplest case of mixing **equal** masses of hot and cold water.

Procedure

Step 1: Begin by marking your three Styrofoam cups equally at about the three-quarter mark. You can do this by pouring water from one container to the next and mark the levels along the inside of each cup.

Step 2: Fill the first cup to the mark with hot water from the pail, and fill the second cup with cold water to the same level. Measure and record the temperature of both cups of water.

Temperature of cold water = 23 °C

Temperature of warm water = 39 °C

Step 3: Predict the temperature of the water when the two cups are combined. Then pour the two cups of water into the liter container, stir the mixture slightly, and record its temperature.

Predicted temperature = 31 °C

Actual temperature of water = 31 °C

If there was a difference between your prediction and your observation, what may have caused it?

Pour the mixture into the sink or waste pail. (Don't be a klutz and pour it back into either of the pails of cold or hot water!) Now you'll investigate what happens when **unequal** amounts of hot and cold



Temperature Mix Lab Manual Physics

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