

# 10.1 Periodic Trends



## 10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

### Section 10.1 – continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	Aluminum
_____	14-18
_____	Transition?

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

#### Column A

- \_\_\_\_\_ 19. It exists in the periodic table.
- \_\_\_\_\_ 20. It has no line periodic table.
- \_\_\_\_\_ 21. Group 1 elements.
- \_\_\_\_\_ 22. Elements that are strong oxidizing agents.
- \_\_\_\_\_ 23. Group 18 elements.

#### Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 24. There are two representative elements of aluminum.
- \_\_\_\_\_ 25. More than three families of the elements in the periodic table are metals.
- \_\_\_\_\_ 26. Group 1A elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 27. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 28. Group 7A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 29. Group 8A elements are inert nonmetallic elements known as noble gases.
- \_\_\_\_\_ 30. Alkaloids have properties of both metals and nonmetals.

# Stoichiometry Study Guide For Content Mastery Key

**John Krenos, Joseph Potenza, Loretta  
Jones, Lynn Koplitz, Thomas Spence**



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