

Chemistry: Atom Unit Study Guide

1.4.10.10 The student reviews the following concepts:

Period _____ Group _____

Atom Unit Study Guide

Definitions: Please define the terms listed in the definitions.

1. **isotopes** _____ atoms of the same element with different numbers of neutrons.

2. **atomic number** _____ the number of protons in an atom.

3. **Mass Number** _____ the number of protons and neutrons in an atom.

4. **Proton** _____ a positively charged particle.

5. **Neutron** _____ a neutral particle.

6. **Electron** _____ a negatively charged particle.

Calculator Usage: Many books, money, calculators, X-rayed discs, markers, pencils, pens, markers, calculators and more than one!

1. When is it ok to use a calculator? _____

2. An atom is made of _____.

3. What subatomic particle in the atom has the least mass? _____

4. The mass number is equal to _____ + _____.

5. Hydrogen has the atomic number _____.

6. Hydrogen is the name given to the atom with the atomic number _____.

Calculations:

1. Determine the number of protons, neutrons, and electrons of $\text{^{19}_{\Lambda}F}$.

2. How many neutrons does the element Beryllium-12 have?

Study Guide The Atom Answers

Robinson

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