

# SEPARATING MIXTURES

	<b>Required materials</b> <b>Method name:</b> Density <ul style="list-style-type: none"> <li>• Objects of different densities</li> <li>• Water</li> </ul>	<b>Describe what happens</b> Place objects of different densities into water. Denser objects sink will follow objects float
	<b>Method name:</b> Magnet <ul style="list-style-type: none"> <li>• Metal (iron)</li> <li>• Nonmetal items</li> <li>• Magnet</li> </ul>	
	<b>Method name:</b> Filter <ul style="list-style-type: none"> <li>• Small particles</li> <li>• Liquid</li> <li>• Filter</li> </ul>	
	<b>Method name:</b> Sieve <ul style="list-style-type: none"> <li>• Small particles</li> <li>• Large particles</li> <li>• Sieve</li> </ul>	

## MIXTURES & SOLUTIONS Study Guide

Word	Definition
Mixtures	A heterogeneous combination of substances that keep their identity
Solution	Mixture created when one substance dissolves in another
Dissolve	Breaks down and becomes part of a liquid.
Solvent	The liquid that dissolves the substance.

Rate each of the following skills:  
 1 = Very difficult to me    2 = Need some help  
 3 = I can almost do this    4 = I can teach a friend

..... I can distinguish between a mixture and solution.  
 ..... I can describe materials that can dissolve in water.  
 ..... I can explain how to separate mixtures and which method to use.

Highlight any phrases that are describing a solution.

- Includes only solid substances
- **Includes 2 liquids**
- **Includes a liquid and a solid**
- Most can be separated
- **Most cannot be separated**
- **Something dissolves**

Circle the mixtures below

# ANSWER KEY INCLUDED

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to dissolve quicker?

- Increase the temperature of the solvent
- Stir the substance in the solvent
- Cut the solute into smaller pieces- this creates more surface area

Salt  
Sugar  
Coffee/cocoa

Name 3 substances that do NOT dissolve in water.

Sand  
Pepper  
Oil



What can you expect to happen when you mix oil and water?

The oil may look like it dissolves at first, but it will separate from the water and either sink or float.

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# Study Guide Mixtures And Solutions Key

**James E. Brady, John R. Holum**



## Study Guide Mixtures And Solutions Key:

*Study Guide to AFCAT 2020 (Air Force Common Admission Test) 6th Edition* Disha Experts, 2019-12-04      Organic Chemistry Study Guide Robert J. Ouellette, J. David Rawn, 2014-11-04 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned by doing It is difficult to learn by rote memorization and true understanding comes only from concentrated reading and working as many problems as possible In fact problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real world problems in the work place Helps readers learn to categorize analyze and solve organic chemistry problems at all levels of difficulty Hundreds of fully worked practice problems all with solutions Key concept summaries for every chapter reinforces core content from the companion book      Chemistry, Student Study Guide John A. Olmsted, Gregory M. Williams, 2005-02-02 100% Pure Chemical Understanding Every morning many of us are energized by a cup of coffee Imagine if you were as energized by understanding the chemistry in your morning cup from the coffee trees which fill red coffee berries with caffeine and a variety of other chemical substances to the feathery crystals formed by the caffeine molecules to the decaffeinating machines which use liquid solvents to remove this stimulant from some of the beans Now that's real chemical understanding Olmsted and Williams Fourth Edition of Chemistry focuses on helping you see and think about the world and even your coffee as a chemist This text helps you understand how chemical phenomena are governed by what happens at the molecular level apply critical thinking skills to chemical concepts and problems and master the basic mathematical techniques needed for quantitative reasoning You'll see the world as chemists do and learn to appreciate the chemical processes all around us A Fourth Edition with a lot of new perks Revisions include a new early energy chapter revised coverage of bonding expanded coverage of intermolecular forces and increased coverage of multiple equilibria including polyprotic acids New pedagogy strengthens students critical thinking and problem solving skills Visual Summaries at the end of each chapter use molecular and diagrammatic visual elements to summarize essential skills concepts equations and terms eGrade Plus provides an integrated suite of teaching and learning resources including a complete online version of the text links between problems and relevant sections in the online text practice quizzes the Visual Tutor Interactive LearningWare problems and lab demos as well as homework management and presentation features for

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