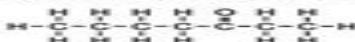


Quiz & Worksheet - Structural Formula

1. Which of the following shows the correct condensed structural formula of a compound that has five carbon (C) atoms, 12 hydrogen (H) atoms and one oxygen (O) atom?

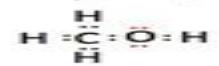
- A $\text{CH}_3\text{CH}(\text{CH}_2)\text{CH}_2\text{C}(=\text{O})\text{H}$
- B $\text{CH}_3\text{CH}(\text{CH}_2)\text{CH}_2\text{C}(=\text{O})\text{H}$
- C $\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}(\text{CH}_3)\text{CH}_2\text{O}$
- D $\text{CH}_3\text{C}(\text{CH}_3)_2\text{CH}_2\text{C}(=\text{O})\text{H}$

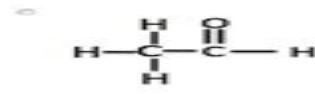
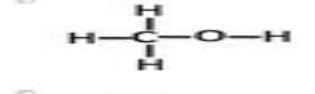
2. Which of the following is the correct skeletal structural formula for the following compound?



- A 
- B 
- C 
- D 

3. What is the correct line-bond structural formula for the compound shown below? (The black dots are bonding electrons between the atoms and the red dots are nonbonding electrons.) Hint: Since this is an organic compound, nonbonding electrons are, in general, not shown in the line-bond structural formula.



- A 
- B 
- C 
- D 

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