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NGSS Regents Chemistry

NOTE PACKET

Unit 12: Nuclear Chemistry

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Nuclear Chemistry

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Unit Vocabulary:

- Alpha particle
- Antineutron annihilation
- Beta particle
- Fission
- Fusion
- Gamma radiation
- Half-life
- Radioisotope
- Transmutation
- Radioactive tracer

Unit Objectives:

Upon completion of this unit you should be able to do the following:

- Explain the stability of an isotope based on the ratio of neutrons and protons in the nucleus.
- Understand that alpha decay model and beta decay are unidirectional and exponentially decaying radiation.
- Calculate the mass percent of the reaction remaining, or the half-life of a radioactive isotope, using the half-life equation.
- Understand the concept of half-life.
- Differentiate between the following emissions based on mass, charge, emitting source, and penetrating power:
 - Alpha
 - Beta
 - Gamma
- Determine the type of decay (alpha, beta, gamma) and write the nuclear equation.
- Distinguish between fission and fusion reactions.
- Distinguish between natural and artificial transmutation.
- Complete nuclear equations and predict missing particles from nuclear equations.
- Understand the changes in energy in a nuclear reaction.
- Be aware of the social uses and real world application of radioactive isotopes:
 - Radioactive dating
 - Tracing chemical and biological processes
 - Industrial measurement
 - Medical science
 - Diagnosis and treatment of diseases

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