

## Chapter 5 The Periodic Table

**WordWise**

Match each definition with the correct term by writing the definition's number in the grid. When you have filled in all the boxes, add up the numbers in each column, row, and the two diagonals. Hint: The sum should be 20 for each case.

**Definitions**

- An arrangement of elements in columns based on a set of properties that repeat from row to row
- A pattern of repeating properties that occurs when atomic numbers are used to arrange elements into groups
- One-twelfth the mass of a carbon-12 atom
- Elements that are good conductors of heat and electric current
- Elements that form a bridge between the elements on the left and right sides of the periodic table
- Elements that are poor conductors of heat and electric current
- Elements with properties that fall between those of metals and nonmetals
- An electron that is in the highest occupied energy level of an atom
- Colorless, odorless, and extremely unreactive gases

			diagonal: _____
nonmetals	periodic table	valence electrons	row _____
metalloids	transition metals	atomic mass unit	row _____
periodic law	noble gas	metals	row _____
_____	_____	_____	diagonal: _____

# Wordwise Answers Guide

**M Tight**

## Wordwise Answers Guide:

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