

Chapter 17 Thermochemistry Study Guide

17.1 – 17.2 Thermochemical Equations

1. Make the following conversions:
 - a. 444 cal to joules = 1.86×10^3 J
 - b. 1.8 kJ to joules = 1.8×10^3 J
 - c. 0.45 kJ to calories = 1.1×10^2 cal
2. Classify each of these processes as endothermic or exothermic:

a. condensing steam – exo	c. evaporating alcohol – endo
b. burning alcohol – exo	d. baking a potato – endo
3. The specific heat capacity for silver is 0.24 J/g°C. Calculate the energy required to raise the temperature of 150.0 g Ag from 273 K to 298 K. Calculate the molar heat capacity of silver.

$$\text{Energy: } q = (150.0)(.24)(298-273) = 9.0 \times 10^2 \text{ J}$$

$$\text{Molar heat capacity: } (0.24 \text{ J/g°C})(107.87 \text{ g/mol Ag}) = 26 \text{ J/mol}$$

4. It takes 585 J of energy to raise the temperature of 125.6 g Hg from 20.0°C to 53.5°C. Calculate the specific heat capacity and the molar heat capacity of Hg.

$$\text{Specific heat capacity: } C = q/m\Delta t = 585/(125.6)(53.5-20) = 0.139 \text{ J/g°C}$$

$$\text{Molar heat capacity: } (0.139 \text{ J/g°C})(200.59 \text{ g/mol Hg}) = 27.9 \text{ J/mol}$$

5. A 46.2-g sample of copper is heated to 95.4°C and then placed in a calorimeter containing 75.0 g water at 19.6°C. The equilibrium temperature in the calorimeter is 21.8°C. Calculate the specific heat capacity of copper; assuming that all the heat lost by the copper is gained by the water.

$$C = 0.203 \text{ J/g°C}$$

6. A 15.0-g sample of nickel metal is heated to 100.0°C and dropped into 55.0 g of water, initially at 23.0°C. Assuming that all the heat lost by the nickel is absorbed by the water; calculate the final temperature of the nickel and the water. The specific heat of nickel is 0.444 J/g°C.

$$15.0(0.444)(100.0 - T_f) = 55.0(4.184)(T_f - 23.0)$$

$$666 - 6.66T_f = 230.12T_f - 5292.76$$

$$5958.76 = 236.78 T_f$$

$$25.2^\circ\text{C} = T_f$$

Thermochemistry Test Study Guide Answers

Clyde R. Metz

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suatu garis yang jaraknya di antara alas diperluas dengan titik sudut adapun garis tinggi segitiga dapat dipakai untuk menghitung luas segitiga yang dirumuskan sebagai setengah dari hasil kali garis tinggi dengan alas
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