

Bonding, Naming, Molecular Structures Study Guide

5 pages + 5-page Answer Key

BONDING, NAMING, AND MOLECULAR SHAPES STUDY GUIDE

1. What are valence electrons, and why are they important in chemical bonding?

2. How does the octet rule influence the behavior of atoms?

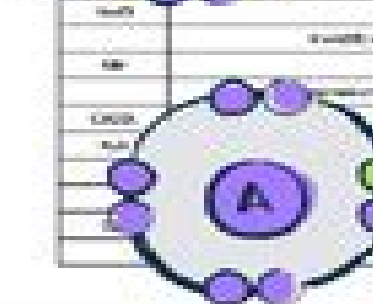
3. What distinguates ionic bonding from covalent bonding, and the types of elements involved?

4. List the common molecular shapes and bond angles.

5. What are resonance structures, and how do they affect the overall properties of a molecule?

6. Draw the Lewis structure for the following compounds:

7. Draw the Lewis structure for the following compounds:



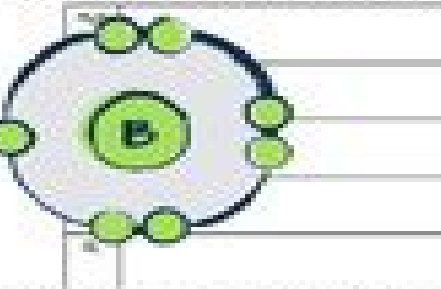
8. What naming rules must be followed when naming ionic compounds?

9. Why is it important to ensure that compounds are correctly named?

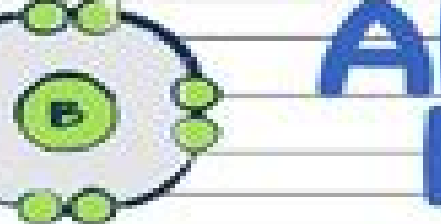
10. Name the following compounds:

NaCl	
KBr	
MgO	
CaS	
LiF	

11. Draw the Lewis structures for the following compounds:



12. Draw the Lewis structures for the following compounds:



Chemistry

Physical Science

13. What are the common molecular shapes and bond angles?

14. What are the common molecular shapes and bond angles?

15. Why is it important to ensure that compounds are correctly named?

16. How will drawing and identifying molecular shapes enhance your understanding of chemistry?

ANSWER KEY INCLUDED

Study Guide Molecular Structures Answers

**T. W. Graham Solomons, Craig B.
Fryhle, Scott A. Snyder**



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