

Name : _____

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Writing and Balancing Equations

Write and balance the following chemical equations.

- When dissolved beryllium chloride reacts with dissolved silver nitrate in water, aqueous beryllium nitrate and silver chloride powder are made.
- When isopropanol burns in oxygen, carbon dioxide, water, and heat are produced.
- When dissolved sodium hydroxide reacts with sulfuric acid, aqueous sodium sulfate, water, and heat are formed.
- When fluorine gas is in contact with calcium metal at high temperatures, calcium fluoride powder is created in an exothermic reaction.
- When sodium metal reacts with iron (II) chloride, they form iron and sodium chloride.
- Zinc and lead (II) nitrate react to form zinc nitrate and lead.
- Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas.
- Sodium phosphate and calcium chloride react to form calcium phosphate and sodium chloride.

Study Guide Writing Balanced Equations

Pamela Mills,Amina El-Ashmawy

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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills,Amina El-Ashmawy,2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Study Material Based On NCERT Science Class- X Dr. Sunita Bhagiya, ,Er. Meera Goyal,2021-11-26

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Becker/Wentworth Becker,Benjamin L. Carroll,1980

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Dresdner, and Mooney William H. Myers,William T. Mooney,1980

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Brady,1990-03-01 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems

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