

Name : \_\_\_\_\_ Date : \_\_\_\_\_

## Writing and Balancing Equations

Write and balance the following chemical equations.

- 1) When dissolved beryllium chloride reacts with dissolved silver nitrate in water, aqueous beryllium nitrate and silver chloride powder are made.
- 2) When isopropanol burns in oxygen, carbon dioxide, water, and heat are produced.
- 3) When dissolved sodium hydroxide reacts with sulfuric acid, aqueous sodium sulfate, water, and heat are formed.
- 4) When fluorine gas is in contact with calcium metal at high temperatures, calcium fluoride powder is created in an exothermic reaction.
- 5) When sodium metal reacts with iron (II) chloride, they form iron and sodium chloride.
- 6) Zinc and lead (II) nitrate react to form zinc nitrate and lead.
- 7) Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas.
- 8) Sodium phosphate and calcium chloride react to form calcium phosphate and sodium chloride.

# Study Guide Writing Balanced Equations

**Pamela Mills,Amina El-Ashmawy**



## Study Guide Writing Balanced Equations:

*The Practice of Chemistry Study Guide & Solutions Manual* Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises      *Study Material Based On NCERT Science Class- X* Dr. Sunita Bhagiya, Er. Meera Goyal, 2021-11-26

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The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems      *Study Guide [to Accompany] General Chemistry* Dixie Goss, 2002      **Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition** John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter      **Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry, 1996 This internationally acclaimed detective series is just the thing for lovers of those Number One Ladies looking for a darker more realistic view of Botswana Sue Baker Publishing News      *Study Guide to Accompany Chemical Principles, Properties, and Reactions* Kenneth L. Henold, 1984      *Basic Inorganic Chemistry* F. Albert Cotton, Geoffrey Wilkinson, Paul L. Gaus, 1995-01-18 Explains the basics of inorganic chemistry with a primary emphasis on facts then uses the student's growing factual knowledge as a foundation for discussing the important principles of periodicity in structure bonding and reactivity New to this updated edition improved treatment of atomic orbitals and properties such as electronegativity novel approaches to the depiction of ionic structures nomenclature for transition metal compounds quantitative approaches to acid base chemistry Wade's rules for boranes and carboranes the chemistry of major new classes of substances including fullerenes and silenes plus a chapter on the inorganic solid state      Introductory

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