

Name _____

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Subatomic Particles and Isotopes Practice

1. Complete the table below.

Element Name	Atomic Number	Mass Number	Number of protons	Number of neutrons	Number of electrons	Isotope Notation
Oxygen		17				
	15	31				
	39			50		
Neon		20				
				10	9	
			79	128		
Iron				30		
				35	29	

2. Answer the following questions:

a) An atom contains 35 protons. What is the element symbol?

b) An atom contains 31 protons, 39 neutrons, and 31 electrons. Identify the mass number of this atom. _____

c) What is the atomic number of bromine? _____

d) How many neutrons in an atom of Hg-201? _____

e) What is the atomic number of Zn-65? _____

f) An atom contains 33 protons. What is the element symbol?

g) An atom contains 17 protons, 18 neutrons, and 17 electrons. Identify the mass number of this atom. _____

Subatomic Particle Gizmo Answers Key

Raffaela Di Napoli

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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