

Chapter 5 Study Guide

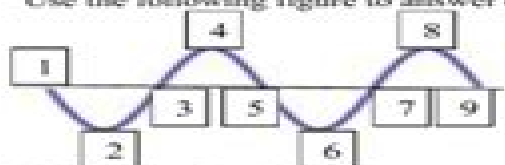
In your textbook, read about the wave nature of light.
Study your notes, charts, tables, and problems that we have completed in class.
Review information in your Science Workbook

Use the following terms below once to complete the passage.

Amplitude	Energy	Frequency	Hertz
Light	Wave	Wavelength	Speed

Electromagnetic radiation is a kind of (1) Energy that behaves like a(n) (2) Wave as it travels through space. (3) Light is one type of electromagnetic radiation. Other examples include x-rays, radio waves and microwaves. All waves can be characterized by their wavelength, frequency and (4) Speed. The shortest distance between equivalent points on a continuous wave is called a(n) (5) Wavelength. The height of a wave from the origin to a crest or from the origin to the trough is the (6) Amplitude. (7) Frequency is the number of waves that pass a given point in one second. The SI unit for frequency is the (8) Hertz, which is equivalent to one wave per second.

Use the following figure to answer questions 9-10



9. Which number(s) represents one wavelength? any of the following: 2-6, 4-8, 1-5, 5-9
10. Which number(s) represent the amplitude? any of the following: 1-2, 3-4, 4-5, 6-5, ect.
11. Circle each of the following things that helped scientist discover the location of the electrons. There may be more than one correct answer.
 - a. Different elements have different emission spectrums.
 - b. deBroglie equation predicted that all moving particles have wavelike characteristics.
 - c. The Heisenberg Uncertainty Principle
 - d. Schrödinger's Wave equation
12. The minimum amount of energy that can be gained or lost by an atom is called Quantum.
13. What was the major contribution to the understanding of the atom:

Einstein: proposed that light has a dual nature (particle and wave). Beam of light made of bundles of energy called photons. Photons with energy above the threshold will cause the photoelectric effect.

Planck: determined that matter gains or loses energy in small amounts called quantum.

Study Guide Light And Quantized Answer

**Katherine J. Denniston, Denniston
Katherine**



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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