

Activity A: Phase changes	Get the Gizmo ready: <ul style="list-style-type: none"> Click Reset () and select Macro view. Set the Water temperature to 10 °C. Set the Ice volume to 0 cm³. 	
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Question: How is temperature related to phase changes?

1. Predict: Based on your prior knowledge, predict the following:

- At what temperature will water change from a liquid to a solid (**freeze**)?
- At what temperature will water change from a solid to a liquid (**melt**)?
- At what temperature will water change from a liquid to a gas (**boil**)?

0

0

100

2. Investigate: Use the Gizmo to explore phase changes. Use the **Add/remove heat energy** slider to control the water temperature. Record your observations in your notes, then answer the questions below:

- At what temperature does water freeze?
- At what temperature does ice melt?
- At what temperature does water boil?

0

0

10

This is the **freezing point**.

This is the **melting point**.

This is the **boiling point**.

3. Observe: Set up the Gizmo to observe freezing. What do you notice about the temperature while the water is in the process of freezing?

Remains unchanged

4. Explore: Use the Gizmo to investigate melting and boiling. Does the temperature change while either of these phase changes is occurring?

No it doesn't

5. Interpret: Select the **GRAPH** tab to see a graph of temperature vs. time. Click the **"-"** button until the whole graph is visible. What does the graph look like during a phase change?

A horizontal line

6. Extend your thinking: Why do you think the temperature does not change much during a phase change? If possible, discuss your answer with your classmates and teacher.

Utilizing the energy to alter the nature of matter

Study Guide Phase Changes Answer Key

Scott A. Bailey, Stephen F. Davis

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