

Matter—Properties and Changes

Lesson 2.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical
density

mass
properties

physical
substance

Matter is anything with (1) MASS and volume. A (2) SUBSTANCE is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) PROPERTIES that can be observed. Substances have both physical and chemical properties. (4) PHYSICAL properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) DENSITY are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) CHEMICAL properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

- | | |
|-------------|---|
| <u>CHEM</u> | 7. Chemical formula H_2O |
| <u>CHEM</u> | 8. Forms green carbonate when exposed to moist air |
| <u>CHEM</u> | 9. Remains unchanged when in the presence of nitrogen |
| <u>PHYS</u> | 10. Colorless |
| <u>PHYS</u> | 11. Solid at normal temperatures and pressures |
| <u>CHEM</u> | 12. Ability to combine with another substance |
| <u>PHYS</u> | 13. Melting point |
| <u>PHYS</u> | 14. Liquid at normal temperatures and pressures |
| <u>PHYS</u> | 15. Boiling point is $100^{\circ}C$ |
| <u>PHYS</u> | 16. Conducts electricity |

Study Guide Properties Of Matter Answers

Arshad Iqbal



Study Guide Properties Of Matter Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03

The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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